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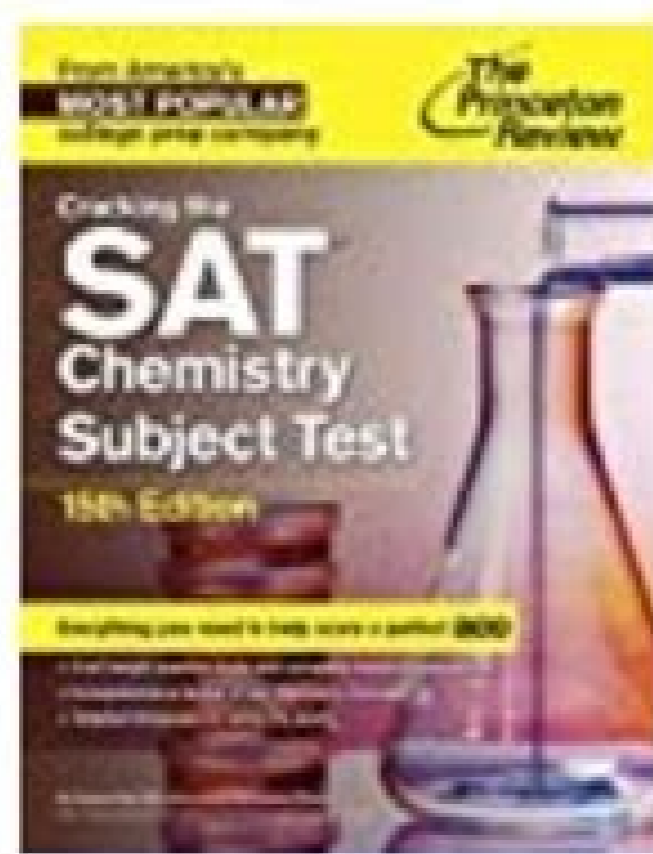
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SAT CHEMISTRY TEST—Continued

34. The correct formula for barium sulfate is
- (A) BaS
(B) Ba₂S
(C) BaSO₄
(D) Ba₂SO₄
(E) BaSO₃
35. The expected products of the decomposition of sodium oxide are
- (A) Na₂ and O₂
(B) Na₂⁺ and O₂²⁻
(C) Na₂⁺ and O₂⁻
(D) Na₂ and O₂⁻
(E) Na₂ and O₂
36. Which of the following sets of conditions best describes a solution of potassium iodide as compared to pure water?
- | Vapor pressure | Freezing point | Boiling point |
|----------------|----------------|---------------|
| (A) higher | higher | higher |
| (B) higher | higher | lower |
| (C) higher | lower | lower |
| (D) lower | lower | higher |
| (E) lower | higher | higher |
37. The elements known for their limited chemical reactivity are the
- (A) Alkali metals
(B) Alkaline earth metals
(C) Transition metals
(D) Halogens
(E) Noble gases
38. Which of the following combinations of 0.20 M aqueous solutions will form a precipitate on mixing?
- I. AgNO₃ and HCl
II. BaCl₂ and MgSO₄
III. KI and Pb(NO₃)₂
- (A) I only
(B) I and II only
(C) I and III only
(D) II and III only
(E) I, II, and III
39. The oxidation number for sulfur is highest in which of the following ions or compounds?
- (A) SO₂
(B) SO₃²⁻
(C) SO₃²⁻
(D) S₂O₈²⁻
(E) HSO₃⁻
40. Given the equation $3X + Y \rightarrow Z$, you react 2 moles of X with 1 mole of Y. Which of the following is true?
- (A) X is the limiting reactant because of its higher molar mass.
(B) X is the limiting reactant because you need 3 moles of X and have 2.
(C) Y is the limiting reactant because you have fewer moles of Y than X.
(D) Y is the limiting reactant because 3 X molecules react with 1 Y molecule.
(E) Neither reactant is limiting.
41. Which of these statements best describes a galvanic cell?
- (A) Reduction occurs at the anode.
(B) Oxidation occurs at the anode.
(C) Energy is required from an external source.
(D) The cathode is negative.
(E) E_{cell} is negative.
42. Which of these half-reactions represents an oxidation?
- I. $\text{Sr}^{2+} \rightarrow \text{Sr}^0$
II. $\text{MnO}_2 \rightarrow \text{Mn}^{2+}$
III. $\text{Cl}_2 \rightarrow \text{ClO}_2$
- (A) I only
(B) II only
(C) III only
(D) I and II only
(E) II and III only
43. Which of the following compounds has more than three isomers?
- (A) C₂H₆
(B) C₃H₈
(C) C₄H₁₀
(D) C₅H₁₂
(E) C₆H₁₄
44. Which of these types of chemical reactions is NOT a redox reaction?
- (A) Decomposition into elements
(B) Single replacement
(C) Double replacement
(D) Combustion
(E) Synthesis of two elements

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Express the nuclear equation: U²³⁸ loses an alpha particle.

$${}_{92}\text{U}^{238} \rightarrow {}_{90}\text{Th}^{234} + {}_2\text{He}^4$$

Sample Flash Card ?

SAT Chemistry Review: Subject Test Basics

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Points	Minutes	Questions
200-800	60	85
		Multiple Choice

more than one test at a sitting

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H																	He
Li	Be											B	C	N	O	F	Ne
Na	Mg											Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac	Rf	Db	Sg	Bh	Ht	Rg	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu

CHEMISTRY TEST

printed on page of the answer sheet that you will see when the Chemistry Test starts is listed as readily as below. When your responses with you to fill in the circles next to the name of the test you are about to take, fill in your answer sheet as shown.

After filling in the circles next to the name of the test you are taking, locate the Background Questions section, which also appears at the top of your answer sheet (as shown above). This is where you will answer the following Background Questions on your answer sheet.

BACKGROUND QUESTIONS

Please answer the four questions below by filling in the appropriate circle in the Background Questions box on your answer sheet.

Question I

Which of the following best describes your preparation in algebra? (If you are taking an algebra course, check a 4.)

Question II

Which of the following best describes your preparation in geometry? (If you are taking an algebra course, check a 4.)

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Question I

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Question II

Which of the following best describes your preparation in geometry? (If you are taking an algebra course, check a 4.)

Our SAT blog is constantly updated to keep up with the content change and testing strategy. This is because the College Board has removed them from their list of officially available practice tests. Use other University Board resources, the College Board has other practice materials available in addition to your full formal practice tests. By taking the practice tests, you can begin to recognize patterns of problems that you lose constantly. You can find a complete list of PSAT practice tests in our guide. Good luck! Our completely free SAT II chemistry practice tests are the perfect way to brush your skills. If so, you should really think about taking the SAT subject test in chemistry to quantify your knowledge of chemistry in a way that is impressive. As you can see from the following proportion of topics that appear in the exam, a large majority of these questions are about chemical principles, but laboratory practices are also fair game: structure of matter (atomic structure, molecular structure, union); 25% States of matter (gases, liquids and solutions); 16% types of reaction (acids and bases, oxidation reduction, precipitation); 14% of stoichiometry (mole concept and chemical equations); 14% Equilibrium system and reaction rates; Thermochemistry 5%; 6% of chemistry Descriptive chemistry; 12% lab; 8% Since there are always eighty-five questions about the SAT subject test in chemistry, you can multiply this total number of questions by the percentage as a decimal to find out roughly how many questions you can expect to face on a given subject on the day of the test. For example, you should be prepared to answer a dozen questions about (0.14 * 85 = 11.9). Each question includes a full explanation of the answer which, in the case of quantitative problems, explains each mathematical step you need to take to get the correct answer. Here are other options you have to get your About quality study materials. Get a preparation book for SAT You must have the same concerns about the preparation books for SAT that you have about the preparation materials for FREE SAT: you want to make sure they are as close as possible from the real business for you to study effectively. Over one of our SAT II chemistry tests and start! See SAT Subject Test In Chemistry Tutors Brianna Certified Tutor Fairfield University, Bachelor of Science, Biology, General. Use free SAT Practice tests (with preparation) free non-official sat practices can be good resources for students who will make much preparation and need supplementary materials. For those who want to prepare with other problems and high quality SAT practices, we offer you some useful alternatives to official practical tests. # 3: Do not use these one-time practicing tests with only 10 official practical tests, it is limited when it comes to optimal practical materials. Many students have problems over time on the SAT exam. Free Official SAT Practice Tests These tests are the only free official tests for the current SAT version. No, the SAT Subject Test in Chemistry is taken by many students for whom chemistry is their favorite subject, and the work of the exam is to combine their actions between SA. As a result, the PSAT Practice Official Tests can be excellent alternatives to SAT practical tests. For problems that are not based on calculations, the response fully explains the choice of the response and how it can eliminate erroneous responses. National University of Ireland in Galway, Master of Science, Neu ... Well, to begin with, just try an individual theme each, very detailed way. Use the PSAT Practice Tests The PSAT is created by the College Board (the creators of the SAT) and looks a lot like the SAT. Calculate each section carefully. These official SAT tests are a good place to start, but you might run out of practice material quickly if you stes eitcarp TAS laiciffo ro rebnum detiml a htW vleitceffe stesT eitcarp TAS laiciffo esU ot woh yasse | snoitanalpxE rewsnA | srewsnA | snoitseuQ :4 tseT eitcarp yasse | snoitanalpxE rewsnA | srewsnA | snoitseuQ :2 tseT eitcarp Tnoitpo taerg a era smaxe eseht neht ,ygetarts dna tnepeganam emit ekil slliks gnikat-tset no gnikrow peek of tnav dna stset eitcarp laiciffo thgie tsrif eht hguorht dekwor ydaerla ev'uoq fl .TAS eht rof perp ot gniyrt sneduts rof ecruser elbailer dna ,eerr ,taerg a eb yawla lliw ralohcSperP nI deknIL yatS ot ralohcSperP esU .stset laiciffo eht fo tuo tsom eht teg ot spets eseht wolloF ,eno suoliverp eht morf sekatsim ruoy deweiver ylhguoroh ev'uoq retfa tset eitcarp tnen eht ot no evom ylno dlouhs uoy taht snaem silt .TAS eht fo srekam eht yb nettirw er'yehet esuaceb evoba detsil airetric eht lla teem draoB egelloC eht morf stset eitcarp TAS lautCa .yltnedifnoc ti ekat dna yrtsimehC ni tseT tcejbuS TAS eht rof flesruoy eraperp nac uoy ,slooT gniarael eerr AAesrotuT ytsraV htiv egdelwonk ruoy gniertslob dna lairetam ssalic ruoy gniweiver yB ,sliaretam perp TAS eerr ot edug evissam ruo htiv stset laiciffonu rof hcraes ruoy tratS .erocs ruoy evorpmi dna sekatsim ruoy morf nrael ot si eitcarp TAS ni ygrene dna emit gniitsevni fo tniop elohw ehT tseT yrevE retfA metroM-tsoP a oD :2# ,ymedaca nahK no smelborp eitcarp devorppa-draoB egelloC esu osia nac uoT .ppa ruoy rof srednow od osia nac maxe desucof elgnis a gnikat fo tneve emit-eno eht ,noitacilpna na no evisserpmi yralucitrap gniraepa eroteb notapicitrap fo sracy ekat netfo notapicitrap strops dna ,maeT etabeD „N,U ledom ,daiipmyO ecneicS ekil setivtica cimedaca ralucitruacitxe ,gnireetnulov sa hcus ,snoitacilpna egelloC fo serutsef ynam elihW .lairetam eht rovo keab gniot retfa tuo meht erugif nac uoy rehtehw ees ot ,jdoirep gniitaw a retfat snoitseuq hguot oder ot tnav nehw thgim uoy ,nalp yduts evinsnetni na evah You will want to make sure to use each and every one of the practical problems in the most effective and efficient way possible. Take one of our many SAT II chemistry practices for a review of the most frequently asked questions. List to point to those 800? You may notice, however, that there is a limit in the amount of FREE Official Resources available. You can consult individual Practice questions on your website for the sections of mathematics, reading, writing and language, and essays. You have probably noticed that the practical test 2 and the practice test 4 are missing in the previous section. What are your other options for SAT practical materials? If you use low-level practical materials, you could be hasty as well as preparing for things that you will not really help you in the SAT, which only you waste your time and it leaves you little prepared, even if you study a lot. Although this is a great resource, keep in mind that there are problems just relying on this service to prepare for SAT. This means that you should do the following: Take each exam directly in a single session, if possible. Is this the chemistry your favorite subject at school, and are looking for a way to give your university requests some definitive feature that allows them to highlight between the crowd? In order to take the maximum advantage of them, it must also recreate real SAT trial conditions as best I can. Even if you can not take all a practice sat directly through, make sure it measures each section appropriately. And do not forget to read the best tips and tricks for Acing the SAT! The test is qualified on a scale of 200 to 800; 800 is a perfect score, while 200 is the worst possible punctuation. Some of our most popular SAT posts include What's next? You will need some high quality resources to study efficiently and, at the same time, make sure to cover each topic property, focusing on those who personally interest you, personally, to be the most challenging. You can do this by working on the problems found in the SAT Subject Test in Chemistry Practice Tests, provided by Varsity Tutors on their Learning Tools website. That said, there are a few things to be careful about when looking for unofficial evidence: Free test preparation materials, by virtue of being free, tend to be of lower quality than paid materials. They may not try the same content in the same way you will see on the real SAT. These tests can be especially useful if you already have a high score and want to delve deeper into what makes official materials different from unofficial ones. When you have identified those topics as the ones you need to focus on, try taking SAT Subject Test in Chemistry Problems organized by topic rather than Practice Tests. View SAT Subject Test in Chemistry Tutors Kevin Certified Tutor University of Pennsylvania, Bachelor of Arts, Chemistry. Chemistry.

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